

Yong-Tao Wang,* Gui-Mei Tang
and Da-Wei QinDepartment of Chemical Engineering, Shandong
Institute of Light Industry, Jinan, Shandong
250100, People's Republic of China

Correspondence e-mail: ceswyt@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.037
 wR factor = 0.108
Data-to-parameter ratio = 10.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2,5-Di-4-pyridyl-1,3,4-oxadiazole–succinic
acid (2/1)In the title structure, $2\text{C}_{12}\text{H}_8\text{N}_4\text{O}\cdot\text{O}C_4\text{H}_6\text{O}_4$, succinic acid molecules lie on crystallographic inversion centers and a pair of $\text{O}-\text{H}\cdots\text{N}$ hydrogen bonds links the three molecules. In addition, $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds link this unit into a two-dimensional framework.

Received 26 September 2005

Accepted 5 October 2005

Online 12 October 2005

Comment

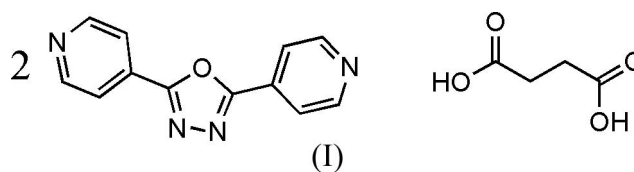
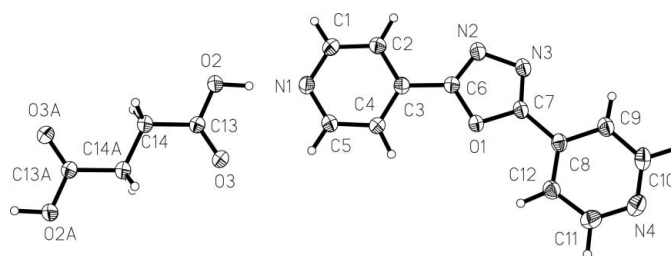
Currently, hydrogen bonding is important in the areas of crystal engineering, supramolecular chemistry, materials science, and biological recognition (Desiraju, 1989; Jeffrey & Saenger, 1991; Holman *et al.*, 2001). Recently, angular dipyridyl-donor basic compounds, such as 2,5-di-4-pyridyl-1,3,4-oxadiazole (bpo), have been used to produce a series of infinite/discrete coordination polymers/supramolecules with interesting structures and properties (Dong *et al.*, 2003; Du *et al.*, 2005a,b, and references therein). In our search to identify the properties of co-crystal materials of fatty diacids with linear/angular base components and to further understand the role of synthons in crystal engineering, we have prepared and determined the crystal structure of the acid–base co-crystal, (I), consisting of bpo and the most typical fatty carboxylic acid, succinic acid (Fig. 1).A view of the title structure is shown in Fig. 1. The asymmetric unit consists of one bpo molecule and half a molecule of succinic acid; a crystallographic inversion center generates the full molecule of succinic acid. In the crystal structure, a two-dimensional framework is formed *via* $\text{O}-\text{H}\cdots\text{N}$, $\text{C}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 2 and Fig. 2).

Figure 1

View of (I), showing displacement ellipsoids at the 30% probability level. H atoms are represented by circles of arbitrary size. Atoms labelled with the suffix A are generated by the symmetry code $(-1 - x, 2 - y, -z)$.

Experimental

A mixture of 2,5-di-4-pyridyl-1,3,4-oxadiazole (112 mg, 0.5 mmol) (Dong *et al.*, 2002) and succinic acid (29 mg, 0.25 mmol) was recrystallized from methanol and water in 70% yield (98 mg), from which a colourless needle-shaped crystal suitable for X-ray diffraction was selected. Analysis found (%): C 59.01, H 3.90, N 19.85, O 16.84; requires (%): C 59.36, H 3.91, N 19.78, O 16.94; IR (KBr, ν cm^{-1}): 2450, 1705, 1612, 1569, 1537, 1413, 1276, 1206, 1011, 836, 741, 723.

Crystal data

$2\text{C}_{12}\text{H}_8\text{N}_4\text{O}\cdot\text{C}_4\text{H}_6\text{O}_4$	$Z = 1$
$M_r = 566.54$	$D_x = 1.444 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 6.4389$ (9) \AA	Cell parameters from 400 reflections
$b = 9.5462$ (13) \AA	$\theta = 7.5\text{--}23.0^\circ$
$c = 11.0415$ (15) \AA	$\mu = 0.11 \text{ mm}^{-1}$
$\alpha = 94.803$ (2) $^\circ$	$T = 293$ (2) K
$\beta = 103.666$ (2) $^\circ$	Block, pale yellow
$\gamma = 96.213$ (2) $^\circ$	$0.43 \times 0.34 \times 0.24 \text{ mm}$
$V = 651.39$ (15) \AA^3	

Data collection

Bruker SMART CCD area-detector diffractometer	2499 independent reflections
φ and ω scans	2277 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.010$
$T_{\text{min}} = 0.956$, $T_{\text{max}} = 0.975$	$\theta_{\text{max}} = 26.0^\circ$
3585 measured reflections	$h = -7 \rightarrow 7$
	$k = -11 \rightarrow 11$
	$l = -10 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0682P)^2 + 0.0637P]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.108$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
2499 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
234 parameters	
All H-atom parameters refined	

Table 1

Selected torsion angles ($^\circ$).

C2—C3—C6—N2	−5.9 (2)	N3—C7—C8—C9	9.7 (2)
-------------	----------	-------------	---------

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
O2—H2B \cdots N1	0.98 (2)	1.70 (2)	2.6741 (15)	174 (2)
C1—H1A \cdots O3 ⁱⁱ	0.94 (2)	2.44 (1)	3.2912 (18)	151 (1)
C10—H10A \cdots N4 ⁱⁱ	0.98 (2)	2.60 (2)	3.433 (2)	143 (2)

Symmetry codes: (i) $x + 1, y, z$; (ii) $-x, -y - 1, -z + 1$.

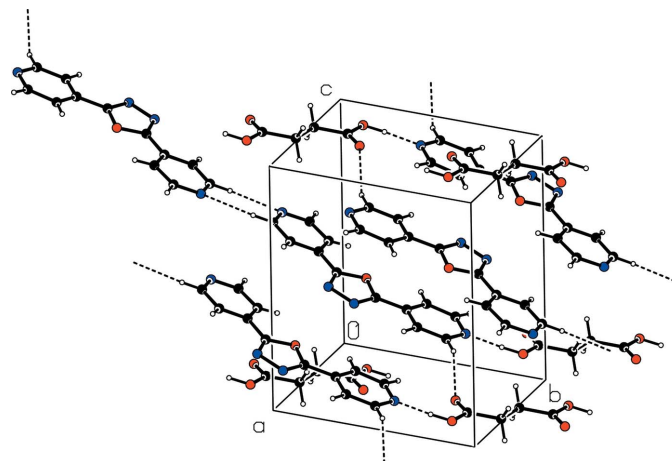


Figure 2

Packing diagram (Spek, 2003), showing hydrogen bonds as dashed lines.

All H atoms were refined independently with isotropic displacement parameters.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 1999); software used to prepare material for publication: SHELXTL.

This work was supported by the Starting Fund of Shandong Institute of Light Industry (to YTW).

References

- Bruker (2001). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Desiraju, G. R. (1989). *Crystal Engineering: The Design of Organic Solids*. New York: Elsevier.
- Dong, Y.-B., Ma, J.-P., Huang, R.-Q., Smith, M. D. & zur Loye, H.-C. (2003). *Inorg. Chem.* **42**, 294–300.
- Dong, Y.-B., Ma, J.-P., Smith, M. D., Huang, R.-Q., Tang, B., Chen, D. & zur Loye, H.-C. (2002). *Solid State Sci.* **4**, 1313–1320.
- Du, M., Zhang, Z.-H. & Zhao, X.-J. (2005a). *Cryst. Growth Des.* **5**, 1119–1208.
- Du, M., Zhang, Z.-H. & Zhao, X.-J. (2005b). *Cryst. Growth Des.* **5**, 1247–1248.
- Holman, K. T., Pivovar, A. M., Swift, J. A. & Ward, M. D. (2001). *Acc. Chem. Res.* **34**, 107–118.
- Jeffrey, G. A. & Saenger, W. (1991). *Hydrogen Bonding in Biological Structures*. Berlin: Springer-Verlag.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97, University of Göttingen, Germany.
- Sheldrick, G. M. (1999). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.